The general equation for acid dissociation is:

$$HA + H_2O \Rightarrow H_3O + A^-$$

The acid dissociation constant, K_a , is given by:

$$K_{a} = \frac{[H_{3}O^{+}][A^{-}]}{[HA]}$$

$$pK_a = -logK_a$$

pH and pOH are defined as:

$$pH = -log[H_3O^+]$$

$$pOH = -log[OH^-]$$

The ionic product of water is given by:

$$K_{\rm w} = [OH^-][H_3O^+] = 10^{-14}$$

$$pK_{w} = pH + pOH = 14$$

Buffer solutions obey the Henderson-Hasselbach equation:

$$pH = pK_a + log \frac{[A^-]}{[HA]}$$

All concentrations are in mol dm⁻³.

1

A solution is made by dissolving 28 g of potassium hydroxide (molar mass = 56) in 100 mL of water. The solution is allowed to cool to 25°C. What is the pH of this solution?

- A. Less than 14
- B. 14
- C. Greater than 14 but less than 15
- D. 15

2

At 25°C the pKa of $H_2PO_4^-$ is 7.21. In order to maintain a system at pH 6.2, the amount of NaH_2PO_4 sodium hydrogen phosphate in the buffer should be:

- A. Twice the amount of Na₂HPO₄
- B. Equal to the amount of Na₂HPO₄
- C. One fifth the amount of Na₂HPO₄
- D. Ten times the amount of Na₂HPO₄

A mole means 6.022 × 10²³ particles.

Relative atomic mass of a sample of an element made of different isotopes:

$$\Delta = m_1$$
 abundance₁ + m_2 abundance₂ + m_3 abundance₃ + ...

abundance₁+ abundance₂+ abundance₃+ ...

Conversion between mass and moles.

relative formula (or molecular) mass $\times M_u$ (g mol⁻¹)

where $M_u = 1 \text{ g mol}^{-1}$

Gas laws: Boyle's law

 $p_1V_1=p_2V_2$

Charles' law

 $\frac{\mathbf{V}_1}{\mathbf{T}_1} = \frac{\mathbf{V}_2}{\mathbf{T}_2}$

Gay-Lussac's law

 $\frac{\mathbf{p}_1}{\mathbf{T}_1} = \frac{\mathbf{p}_2}{\mathbf{T}_2}$

Avogadro's law

 $\frac{V}{n}$

Ideal Gas law

pV = nRT

Kinetic Gas equation

 $pV = \frac{mnC_{RMS}}{3}$

Graham's law of diffusion

 $Rate_1 \propto \frac{1}{\sqrt{M}}$

Graham's law of diffusion 2

 $\frac{Rate_1}{Rate_2} = \sqrt{\frac{M_2}{M_1}}$

Where m = mass, p = pressure, V = volume, n = number of molecules, T = temperature, R = the gas constant, C_{RMS} = root mean squared velocity, M_x is the molecular mass of gas x

1

At standard temperature and pressure, fluorine gas has a density of 1.696 gL⁻¹ and chlorine gas has a density of 3.214 gL⁻¹. Which of these is the best estimate of the relative rates of diffusion of fluorine and chlorine?

- A. $Rate_{CI} = \frac{1}{\sqrt{2}} Rate_{F}$
- B. $Rate_{CI} = \frac{1}{2} Rate_{F}$
- C. Rate_{Cl} = $\sqrt{2}$ Rate_F
- D. $Rate_{Cl} = 2 Rate_{F}$

2

An ideal gas is heated from 400 K to 900 K. What will happen the the root mean squared velocity (C_{RMS})?

- A. The root mean squared velocity will not change
- B. The root mean squared velocity will increase by a factor of $\frac{4}{9}$
- C. The root mean squared velocity will increase by a factor of $\frac{3}{2}$
- D. The root mean squared velocity will increase by a factor of $\sqrt{3} \times \frac{3}{2}$

$$aA + bB \Rightarrow cC + dD$$

When all species are aqueous, the equilibrium constant, $K_{\rm C}$, can be expressed as:

$$K_{c} = \frac{[C]^{c}[D]^{d}}{[A]^{a}[B]^{b}}$$

When all species are gaseous, the equilibrium constant, K_P , can be expressed as:

$$K_{P} = \frac{(p_{C})^{c} (p_{D})^{d}}{(p_{A})^{a} (p_{B})^{b}}$$

where
$$p_A = x_A \times p_{tot}$$
; $x_A = \frac{\text{no. moles of gas A}}{\text{total no. of moles of gas}}$; $x_{tot} = x_A + x_B + x_C + x_D = 1$

Dalton's law:

$$p_{tot} + p_A + p_B + p_C + p_D$$

Raoult's law: the vapour pressure of a solution, p, is equal to the vapour pressure of the pure solvent, p_{solv}° , multiplied by the mole fraction of the pure solvent, x_{solv} :

$$p = x_{solv} \times p_{solv}^{\circ}$$

Raoult's law applied to a mixture of two volatile liquids, A and B, at equilibrium:

$$p = x_A \times p_A^\circ + x_B + p_B^\circ$$

Henry's law: for real solutions at low concentrations:

$$p_{\rm C} = x_{\rm C} \times K_{\rm H}$$

Note: K_C and K_P = equilibrium constants, [X] = concentration of X, p_A = partial pressure of A, x_A = mole fraction of A, p_{tot} = total pressure, K_H is Henry's constant.

1

600 g of ethane (Mw = 30 gmol⁻¹) is mixed with 220 g propane (Mw = 44 gmol⁻¹). The total pressure of the system is 120 mm Hg. What is the partial pressure of propane?

- A. 24 mm Hg
- B. 30 mm Hg
- C. 90 mm Hg
- D. 96 mm Hg

2

A mixture of water and acetone boils at 80°C under atmospheric pressure (1 atm) The vapour pressures of acetone and water under these conditions are approximately 2 atm and 0.5 atm respectively. What is the mole fraction of water?

- A. $\frac{1}{4}$
- B. $\frac{1}{3}$
- C. $\frac{2}{3}$
- D. $\frac{3}{4}$

Rates

For any reaction, the rate is given by:

Rate =
$$k[A]^a [B]^b [C]^c...$$

where k is the rate constant; A,B, and C are reactants which appear in the rate determining step; a,b,c and are their stoichiometric coefficients.

The rate constant varies with temperature according to the Arrhenius equation:

$$k = Ae^{\frac{-E_s}{RT}}$$

$$\ln(k) = \ln(A) - \frac{E_a}{RT}$$

where A is the Arrhenius constant, E_a is the activation energy, R is the gas constant and T is the temperature in kelvin.

For enzyme reactions of the form:

$$E + S \underset{k_{-}}{\rightleftharpoons} ES \xrightarrow{k_{cat}} E + P$$

The Michaelis-Menten equation applies:

$$V_0 = \frac{d[P]}{dt} = V_{\text{max}} \frac{[S]}{K_M + [S]}$$

where V_0 = initial rate; V_{max} = maximum rate; and

$$K_{\rm M} = \frac{k_{\rm r} + k_{\rm cat}}{k_{\rm f}}$$

Thermodynamics

The heat produced from a reaction can be calculated using:

$$Q = mc\Delta T$$

where Q = heat produced; m = mass of reactants; ΔT = temperature change; c = specific heat capacity.

Hess' law states that the enthalpy change of reaction, ΔH_{net} , is independent of the route taken:

$$\Delta H_{\text{net}} = \sum \Delta H_r$$

Entropy is a measure of disorder. The entropy change for a reaction, ΔS_r , is given by:

$$\Delta S_r = \sum S^\circ \text{ (products)} - \sum S^\circ \text{ (reactants)}$$

For an internally reversible, isothermal process:

$$\Delta S_r = \frac{Q}{T}$$

Gibbs free energy change, ΔG°, is defined as:

$$\Delta G^{\circ} = \Delta H^{\circ} - T\Delta S^{\circ}$$

 $\Delta G^{\circ} = -RT \ln(K)$

where K = an equilibrium constant.

1

For an enzyme with $K_{\rm M}$ = 0.75 mmol dm⁻³ and a $V_{\rm max}$ of 400 mmol s⁻¹, what substrate concentration will result in a rate which is $\frac{1}{4}$ of the maximum?

- A. $\frac{1}{4}$ mmol dm⁻³
- B. $\frac{1}{2}$ mmol dm⁻³
- C. 4 mmol dm⁻³
- D. $5\frac{1}{3}$ mmol dm⁻³

2

The hydrophobic effect is when a non polar molecule, such as a protein, aggregates on addition to water. Which of the following is true regarding the thermodynamics of aggregation?

- A. ΔG is positive
- B. ΔS° is negative
- C. It is an endothermic reaction
- D. ΔG° is independent of temperature

Equations of Linear Motion:

When there is no acceleration (a = 0):

$$s = vt$$

At constant acceleration:

$$v = u + at;$$

$$s = ut + \frac{1}{2} at^2;$$

$$v = u + at;$$
 $s = ut + \frac{1}{2} at^2;$ $s = \frac{1}{2} (u + v)t;$ $v^2 = u^2 + 2as$

$$v^2 = u^2 + 2as$$

Momentum and kinetic energy:

$$\rho$$
 = mv; $E_K = \frac{1}{2}mv^2$

Remember, velocity and acceleration are the first and second derivatives of displacement

Equations of Angular Motion:

When there is no acceleration (a = 0):

$$\omega = \frac{\Delta \theta}{\Delta t}; \qquad \theta = \theta_0 + \omega t$$

$$\theta = \theta_0 + \omega t$$

At constant acceleration: $\omega = \omega_0 + \alpha t$; $\Delta \theta = \omega_0 t + \frac{1}{2} \alpha t^2$; $\omega^2 = \omega_0^2 + 2\alpha \Delta \theta$

$$\omega = \omega_0 + \alpha t$$

$$\Delta\theta = \omega_0 t + \frac{1}{2} \alpha t^2$$

$$\omega^2 = \omega_0^2 + 2\alpha\Delta\theta$$

Angluar momentum, rotational kinetic energy and angular speed:

$$E_K = \frac{1}{2} I = 2$$

$$L = I\omega;$$
 $E_K = \frac{1}{2}I = 2;$ $\omega = 2\pi f = \frac{2\pi}{t}$

Magnitude of angular terms (v, a, and F are perpendicular to r):

$$\omega = \frac{v}{r}$$

$$\alpha = \frac{a}{r}$$

$$\omega = \frac{v}{r}$$
; $\alpha = \frac{a}{r}$; $L = mvr$; $\tau = Fr$; $l = mr^2$

Where $v = final \ velocity$, $u = initial \ velocity$, $a = linear \ acceleration$, s = displacement, t = time, $\rho = momentum$, m = mass, E_K = kinetic energy, ω = angular velocity, ω_0 = angular velocity at t = 0, $\Delta\theta$ = angular displacement, L = angular momentum, I = moment of inertia, f = frequency, r = radius, τ = torque and F = force

An object accelerating at a rate of 2 ms⁻² has an initial velocity of 10 ms⁻¹ and a final velocity of 15 ms⁻¹. What distance does the object travel (answers given to 2 s. f.)?

- A. 5.0 m
- B. 23 m
- C. 31 m
- D. 38 m

A gramophone reads music from a record at a constant rate. When the gramophone switches from playing a song at the inner edge of a record to the outer edge the frequency drops from 10 Hz to 4 Hz. What is the effect on the rotational kinetic energy of the record?

- A. It remains constant
- B. It decreases by 40%
- C. It decreases by 60%
- D. It decreases by 84%

Force:

$$F = ma$$
$$F = \frac{\Delta (mv)}{\Delta t}$$

Impulse:

$$I = F\Delta t = \Delta(mv)$$

Moment:

$$moment = F d cos \theta$$

Energy, Work and Power

$$E_{\rm K} = \frac{1}{2} \, m v^2$$

$$\Delta E_P = mg\Delta h$$

At constant force:

$$W = \int F \cdot ds$$

$$P = \frac{\Delta W}{\Delta t} = Fv$$

Force between two masses:

$$F = \frac{Gm_1m_2}{r^2}$$

The gravitational field strength:

$$F = mg;$$

$$g = \frac{GM}{r^2}$$

Gravitational potential:

$$E_{\rm P} = -\frac{GmM}{r}$$
;

$$g = -\frac{\Delta E_{P}}{\Delta r}$$

Where F = force, a = acceleration, I = impulse, v = velocity, t = time, m = mass, E_K = kinetic energy, E_P = gravitational potential energy, h = height, g = gravitational acceleration, W = work done, s = displacement, θ = angle between applied force and displacement, P = power, r = distance,

G = gravitational constant, M = mass of Earth.

1

The mass of the Earth is approximately 100 times larger than the mass of the Moon. There is a point, P, between the Moon and the Earth at which the gravitational field strength is zero. Which of the following is true?

- A. P is equidistant from the Earth and the Moon
- B. P is 10 times closer to the Moon than the Earth
- C. P is 100 times closer to the Moon than the Earth
- D. P is 10,000 times closer to the Moon than the Earth

2

Which of the following statements is not true of the velocity needed to escape the gravitational pull of the Earth?

- A. The velocity is independent of the mass of the escaping object
- B. The velocity is dependent on the gravitational constant of the Earth
- C. The velocity will be affected by atmospheric friction
- D. The velocity is inversely proportional to the squared distance between the object and the Earth's surface

Stress, Strain and Hooke's Law

Stress:
$$\sigma = \frac{F_n}{A}$$
; $\tau = \frac{F_p}{A}$

Strain:
$$\varepsilon = \frac{\Delta l}{l_0} = \frac{\sigma}{E}$$

$$\gamma = \frac{s}{d} = \frac{\tau}{G}$$

Hooke's law:
$$k = E \frac{A}{l_0}$$
; $F = -kx$

Where σ = normal stress, F_n = normal force perpendicular to area, A = area, T = shear stress, F_p = shear force in the plane of the area, ε = strain (unitless), ΔI = change of length, I_0 = initial length, E = Young's modulus, V = shear strain (unitless), V = displacement of the faces, V = distance between faces, V = Shear modulus of elasticity, V = spring constant, V = force and V = length of extension.

Fluid Dynamics

$$\rho = \frac{m}{v}$$

$$p = \frac{F}{A}$$

$$p = p_0 + \rho g h$$

$$p = \frac{1}{2} \rho v^2 + \rho g h = constant$$

$$A_1v_1 = A_2v_2 = constant$$

Where ρ = density, m = mass, V = volume, p = pressure, p_0 = initial pressure, F = force, A = area, g = gravitational acceleration, h = depth of fluid, v = fluid velocity.

1

A force of 10 kN is applied to the end of a circular steel rod with a modulus of elasticity of 200 ×10⁹ Nm⁻², a length of 2 m and a diameter of 10 mm. What is the change, in metres, of the length of the rod?

A.
$$\frac{1}{\pi} \times 10^{-11}$$

B.
$$\frac{1}{25\pi} \times 10^{-8}$$

C.
$$\frac{1}{1000\pi}$$

D.
$$\frac{1}{250\pi}$$

2

Water flows through a horizontal pipe at a velocity of 5 ms⁻¹. The cross-sectional area of the pipe then reduces by a factor of two. Which of these statements is true?

- A. Both the fluid pressure and fluid velocity increase.
- B. Both the fluid pressure and the volume flow rate decrease.
- C. The fluid pressure increases but the fluid velocity decreases.
- D. The fluid pressure decreases and the volume flow rate remains constant.

Wave Phenomena

$$v = f\lambda$$

$$T = \frac{t}{n}; \quad f = \frac{1}{T}$$

$$y(x,t) = A \sin(kx - \omega t)$$

$$\lambda = \frac{2\pi}{k}; \quad v = \lambda f = \frac{\omega}{k}; \quad I \propto A^2$$

Where v = wave velocity, f = frequency, λ = wavelength, T = period, t = time, n = number of cycles, A = amplitude, ω = angular frequency, k = angular wave number, l = intensity, x = co-ordinate along the direction of the wave.

Sound

Doppler effect: $f_o = f_s \left(\frac{v \pm v_o}{v \pm v_s} \right)$

Intensity of sound: $\beta = 10\log_{10}\left(\frac{I}{I_{10}}\right)$

Superposition: $I_{max} = |A_1 + A_2|^2$; $I_{min} = |A_1 - A_2|^2$

Two sinusoidal waves with same amplitude, angular frequency and angular wave number but different phase result in:

 $y = A \sin\left(kx - \omega t + \frac{\Phi}{2}\right)$

Interference of coherent waves where amplitude and angular frequency are equal:

$$A = 2A_1 \cos\left(\frac{\Phi}{2}\right); \quad I = 4I_1 \left(\cos\left(\frac{\Phi}{2}\right)\right)^2$$

If Φ = 0, 2π , 4π , ... then the interference is constructive, if Φ = π , 3π , 5π ... then the interference is destructive.

Where Φ = phase, f_o = observed frequency, f_s = source frequency, v_s = wave velocity, v_s = source velocity, v_o = observer velocity, θ = intensity level in decibel, θ = intensity, θ = intensity of a reference signal.

1

Noise cancelling headphones produce sound waves to destructively interfere with detected external sound waves. What is the phase and intensity of the speaker relative to the external sound?

- A. $\phi = 0$ and $I_{speaker} > I_{external}$
- B. $\phi = \pi$ and $I_{speaker} = I_{external}$
- C. $\phi = 2\pi$ and $I_{speaker} = I_{external}$
- D. $\phi = 2\pi$ and $I_{speaker} > I_{external}$

2

The speed of sound on Mars is approximately 240 ms⁻¹. Two space buggies are both moving at a speed of 40 ms⁻¹ towards each other. If one buggy sends a signal with a frequency of 100 Hz, what wavelength will be perceived by the other buggy?

- A. 100 cm
- B. 140 cm
- C. 171 cm
- D. 338 cm

Wave Phenomena

$$v = f\lambda$$

$$T = \frac{t}{n}; \quad f = \frac{1}{\tau}$$

$$y(x,t) = A \sin(2\pi \left(\frac{x}{\lambda} - ft\right) + \Phi)$$

$$y(x,t) = A \sin(kx - \omega t + \Phi)$$

Where v = wave velocity, f = frequency, $\lambda =$ wavelength, T = period, t = time, n = number of cycles, A = amplitude, $\Phi =$ phase, $\omega =$ angular frequency, k = angular wave number.

Light

The law of reflection: $\theta_i = \theta_i$

The index of refraction: $n = \frac{c}{v_{\text{apparent}}}$

Snell's law $n_1 \sin \theta_1 = n_2 \sin \theta_2$

Critical angle for total internal reflection: $\theta_c = \sin^{-1}\left(\frac{n_2}{n_2}\right)$

Lateral magnification: $M = \frac{h'}{h}$

Lens maker's formula and the thin lens equation: $\frac{1}{f} = \frac{n_2 - n_1}{n_1} \left(\frac{1}{r_1} - \frac{1}{r_2} \right); \quad \frac{1}{u} + \frac{1}{v} = \frac{1}{f}$

where θ_i = angle of incidence, θ_r = angle of reflection, n = index of refraction, c = speed of light, v = velocity, M = magnification, h = actual height of object, h' = height of image, f = focal length, r_x = radius of curvature, u = object distance, v = image distance.

Young's Double Slit Experiment

$$\lambda = \frac{\alpha x}{D}$$

where λ = wavelength of source, α = distance between centres of the slit, x = fringe width, D = distance from double slit to the screen.

The Diffraction Grating

$$d \sin \theta = n\lambda; \quad d = \frac{1}{N}$$

where d = spacing between adjacent slits, θ = angular separation between maxima, n = order of maxima, λ = wavelength of source.

1

With reference to the thin lens formula, which of the following will result in a virtual, upright image 60 cm from the lens?

- A. An object distance of 20 cm and a focal length of 15 cm
- B. An object distance of 65 cm and a focal length of 5 cm
- C. An object distance of 15 cm and a focal length of 20 cm
- D. An object distance of 30 cm and a focal length of 30 cm

2

When Young's double slit experiment is conducted using orange light of wavelength 6.0×10^{-7} m a fringe separation of 2.0 mm is obtained. The light source is changed to violet light of wavelength 4.0×10^{-7} m. What is the new fringe separation?

- A. 0.66 mm
- B. 1.3 mm
- C. 2.6 mm
- D. 3.0 mm

Coulomb's Law

$$F = \frac{1}{4\pi\epsilon_0} \frac{Q_1 Q_2}{r^2} = k \frac{Q_1 Q_2}{r^2}$$

Force on a charge and work done:

$$F = EQ;$$
 $\Delta W = Q\Delta V$

Field strength (uniform field) and radial field strength:

$$E = \frac{V}{d}; \qquad E = \frac{1}{4\pi\varepsilon_0} \frac{Q}{r^2}$$

Electric potential:

$$V = \frac{1}{4\pi\varepsilon_0} \frac{Q}{r^2}$$

Capacitance:

$$C = \frac{Q}{V}; \qquad E = \frac{1}{2}QV$$

$$Q = Q_0(1 - e \frac{-t}{RC})$$

Where F = force, $\varepsilon_0 =$ permittivity of free space (constant), Q = charge, r = distance, k =the Coulomb constant, E =electric field strength, V =potential difference, W =work done, C = capacitance, R = resistance, t = time.

$$F = BIl;$$
 $F = BQv$

$$\Phi = BA;$$
 $\Phi = BA \cos \theta;$ $\varepsilon = N \frac{\Delta \Phi}{\Delta t}$

$$\varepsilon = N \frac{\Delta \Phi}{\Delta t}$$

Rotating coil:

 $\varepsilon = BAN \omega \sin \omega t$

Alternating current:

$$I_{rms} = \frac{I_0}{\sqrt{2}}$$
 ; $V_{rms} = \frac{V_0}{\sqrt{2}}$

Transformers:

$$\frac{N_s}{N_D} = \frac{V_s}{V_D};$$

$$\frac{N_s}{N_p} = \frac{V_s}{V_p}$$
; efficiency = $\frac{I_s V_s}{I_p V_p}$

Where F = force, B = magnetic field strength, l = length, Q = charge, v = velocity, Φ = magnetic flux, A = area, θ = angle between a perpendicular vector to the area and the magnetic field, ε = induced voltage, N = number of turns on coil, t = time, ω = angular momentum, V_{rms} = root mean squared voltage, V_0 = initial voltage, I_{rms} = root mean squared current, I_0 = initial current, N_p = number of turns on primary coil,

 N_s = number of turns on secondary coil, V_p = voltage across primary coils,

 V_s = voltage across secondary coils, I_p = current (primary coil), I_s = current (secondary coil).

What is the direction of the electric field which accelerates a negative charge in a westerly direction?

- A. North
- B. South
- C. East
- D. West

A surface carries a uniform magnetic field of flux density 3 T. A horizontal straight wire of length 0.1 m carrying a current of 0.2 A floats above the surface. What is the mass of the wire? (recall F = mg)

- A. 0.6 g
- B. 6 q
- C. 0.06 kg
- D. 0.6 kg

Ohm's Law:

$$I = \frac{\Delta Q}{\Delta t}$$
; $V = \frac{W}{Q}$; $R = \frac{V}{I}$

Resistivity

$$R = \frac{\rho L}{A}$$

Resistors in Series

$$R_T = R_1 + R_2 + R_3 + \dots$$

Resistors in Parallel

$$\frac{1}{R_T} = \frac{1}{R_1} + \frac{1}{R_2} + \frac{1}{R_3} + \dots$$

Kirchhoff's Current Law:

$$\sum_{k=1}^{n} I_k = O$$

Kirchhoff's Voltage Law:

$$\sum_{k=1}^{n} V_k = O$$

Power

$$P = VI = I^2R = \frac{V^2}{R}$$

EMF

$$\varepsilon = \frac{E}{Q} = I(R + r)$$

Where I = current, Q = charge, t = time, V = potential difference, W = work done, R = electrical resistance, ρ = specific resistivity, L = length, A = area, ε = electromotive force, E = Energy in the circuit, r = internal resistance of the cell.

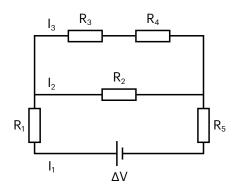
1

Five lamps are connected in parallel to a 12 V battery. The resistance of each lamp is 6 Ω . The current flowing through the battery and the power consumed by each lamp is:

- A. 10 A and 24 W
- B. 10 A and 600 W
- C. 2 A and 24 W
- D. 2 A and 48 W

2

Consider the following circuit. Which of the following is the correct expression with regards to the following circuit?



- A. $I_2R_2 + I_3(R_3 + R_4) = 0$
- B. $-I_2R_2 + I_3(R_3 + R_4) = 0$
- C. $\Delta V I_1(R_1 + R_5) + I_3(R_3 + R_4) = 0$
- D. $\Delta V + I_1(R_1 + R_5) I_3(R_3 + R_4) = 0$

Photons and **Energy Levels:** **Photon energy:**

 $E = hf = \frac{hc}{\lambda}$

Photoelectricity and energy levels:

 $Ek_{(max)} = hf - \Phi;$ $hf = E_1 - E_2$

De Broglie wavelength:

$$\lambda = \frac{h}{p} = \frac{h}{mv}$$

Where E = energy, h = Planck's constant, f = frequency, c = speed of light, λ = photon wavelength, $E_{k(max)}$ = maximum kinetic energy of ejected electron, Φ = threshold energy, p = linear momentum, m = mass, v = velocity.

Radiation:

Inverse square law (y radiation):

 $I \propto \frac{1}{v^2}$

Radioactive decay:

 $A = -\frac{\Delta N}{\Delta t} = \lambda N;$ $N = N_0 e^{-\lambda t};$

Half-lives:

 $T\frac{1}{2} = \frac{\ln 2}{\lambda}; \qquad T\frac{1}{2} = \frac{\ln 2}{\lambda};$

Where I = intensity, x = distance from source, A = activity, N = number of particles, t = time, λ = decay constant, N₀ = number of particles at t=0, T_2^1 = half-life.

X-Rays and **Ultrasound:**

 $I = I_0 e^{-\mu x}; \quad \mu_m = \frac{\mu}{\rho}$

 $Z = \rho c;$ $\frac{I_r}{I_l} = \left(\frac{Z_2 - Z_1}{Z_2 + Z_1}\right)^2$

Where I = transmitted intensity, I_0 = incident intensity, μ = linear attenuation coefficient, $\mu_{\rm m}$ = mass attenuation coefficient, ρ = density, x = path length, Z = characteristic impedance, c = speed of sound in the material, I_r = intensity of reflected wave, I_i = intensity of incident wave, Z_1 = acoustic impedance of material 1, Z_2 = acoustic impedance of material 2.

1

A sample contains two radioactive nuclei, A and B. The half lives of A and B are 3 and 9 minutes respectively. After 18 minutes, the sample contains an equal amount of A and B. What was the initial ratio of A:B?

- A. 1:1
- B. 4:1
- C. 16:1
- D. 32:1

An ultrasound wave is directed at a tissue boundary. What percentage reflection would be expected if $Z_1 = 3 Z_2$?

- A. 0.25 %
- B. 25 %
- C. $25\sqrt{Z_2}$ %
- D. $25 \times \frac{1}{\sqrt{7}}\%$